

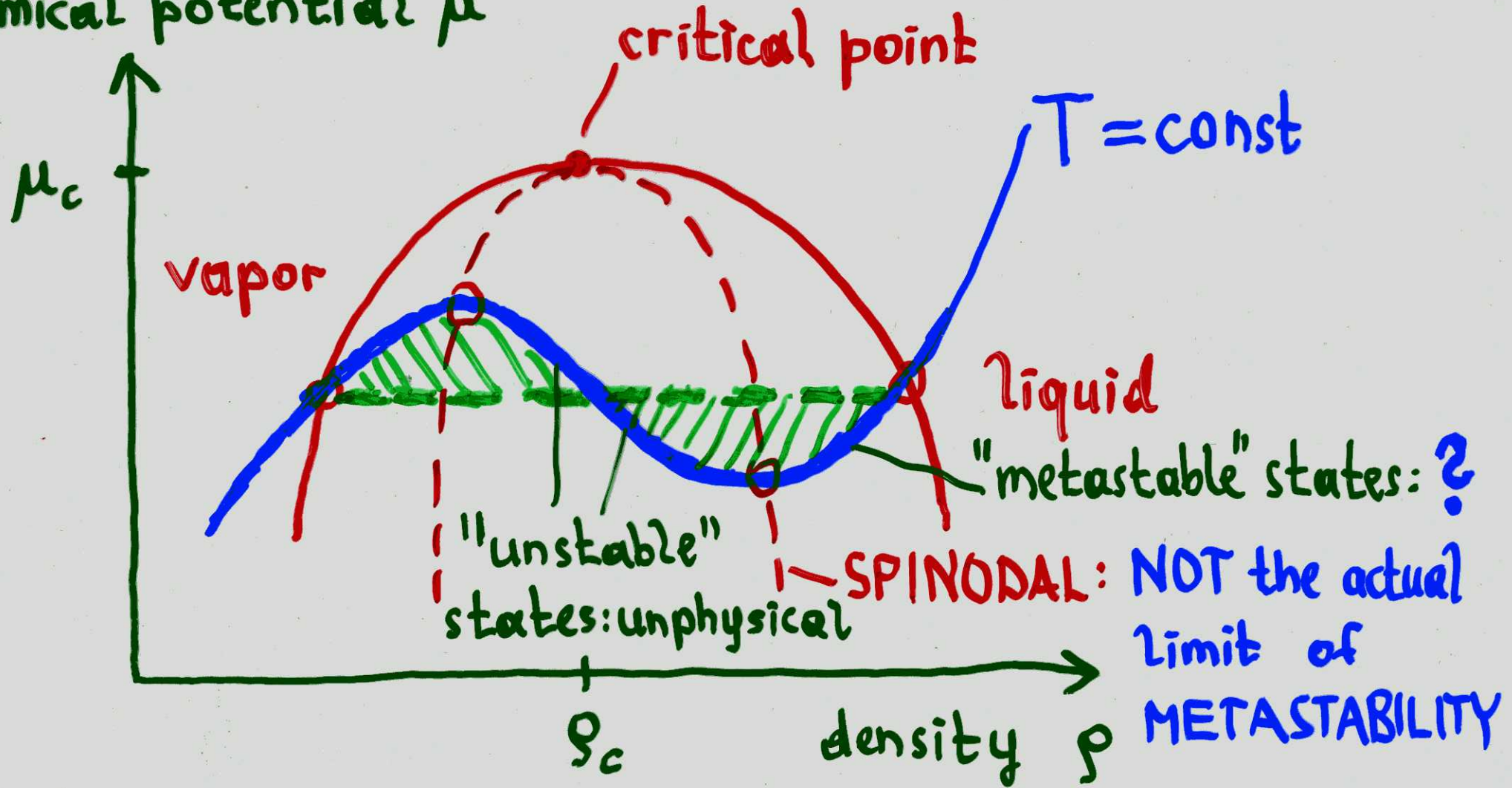
NUCLEATION PROCESSES AND THEIR STUDY BY SIMULATIONS

Kurt Binder

Coworkers: D. Stauffer, H. Müller-Krumbhaar,
H. Furukawa, D.W. Heermann, L.G. MacDowell,
M. Müller, P. Virnau, L. Yelash

van der Waals theory: MEAN FIELD!

chemical potential μ

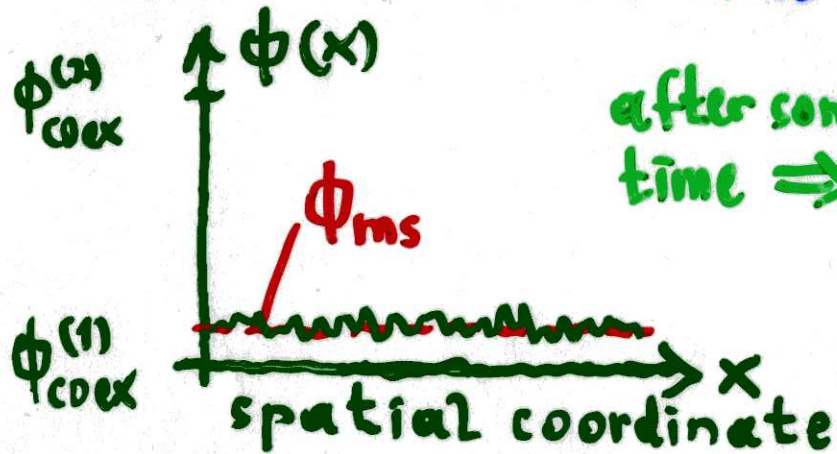
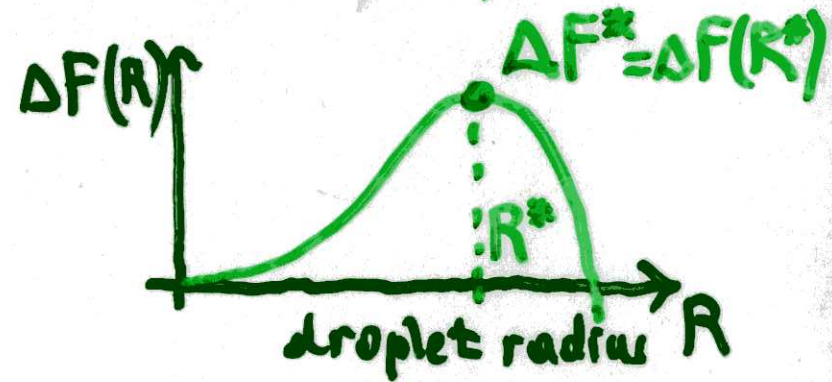


PHASE COEXISTENCE ?

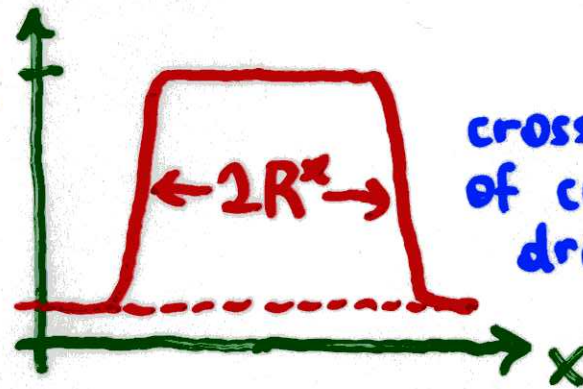
Maxwell construction is ad hoc!

MECHANISMS for the INITIAL STAGES of PHASE SEPARATION KINETICS (mean field theory: CAHN-HILLIARD 1959)

in between coexistence curve and spinodal;
NUCLEATION BARRIER must be overcome.
CRITICAL DROPLET must be formed



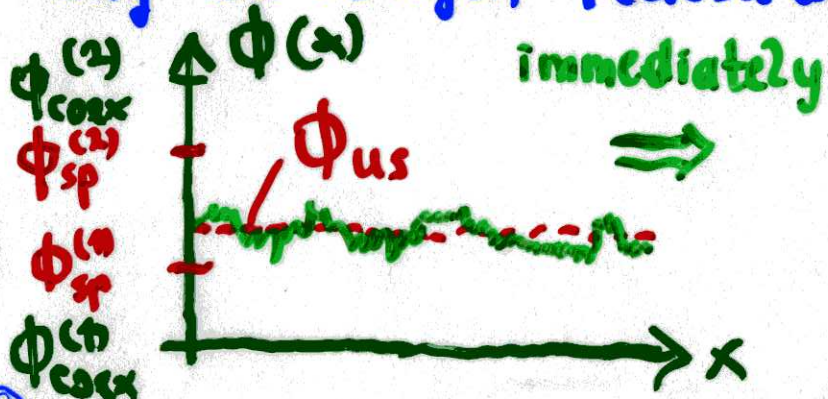
after some time \Rightarrow



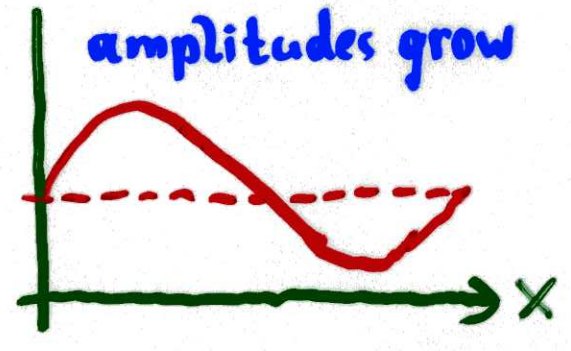
cross section of critical droplet \Rightarrow

droplet growth

inside the spinodal region: $\phi_{sp}^{(1)} < \phi < \phi_{sp}^{(2)}$ **SPINODAL DECOMPOSITION**
 long wavelength fluctuations get spontaneously **AMPLIFIED**



wave packet of unstable waves \Rightarrow

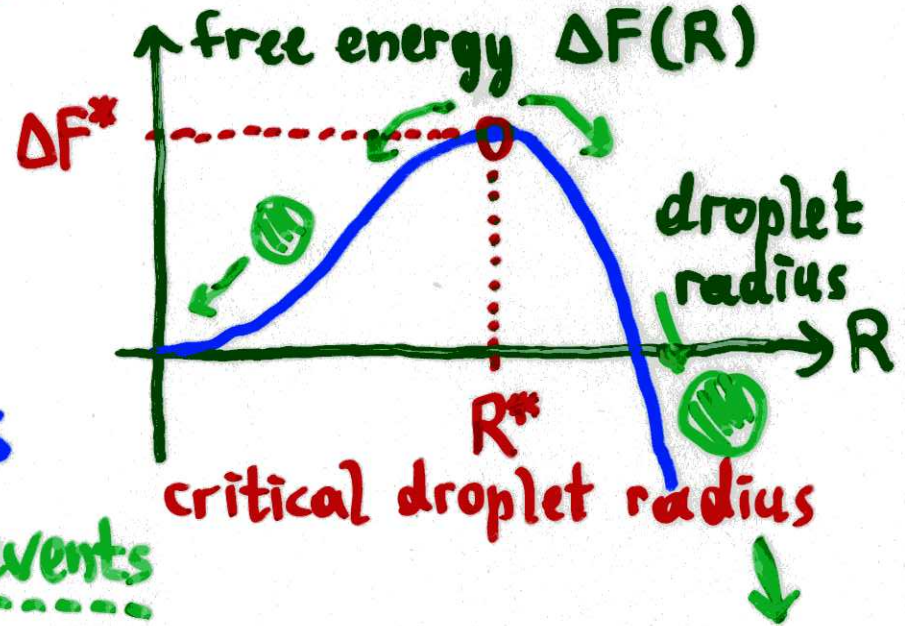


HOMOGENEOUS VERSUS HETEROGENEOUS NUCLEATION

homogeneous nucleation:

a "droplet" of the new (stable) phase forms from the old (metastable) phase by SPONTANEOUS THERMAL FLUCTUATIONS

- high free energy barrier $\Delta F^* \Rightarrow$ rare events
- R^* nanoscopic: direct observation DIFFICULT



heterogeneous nucleation

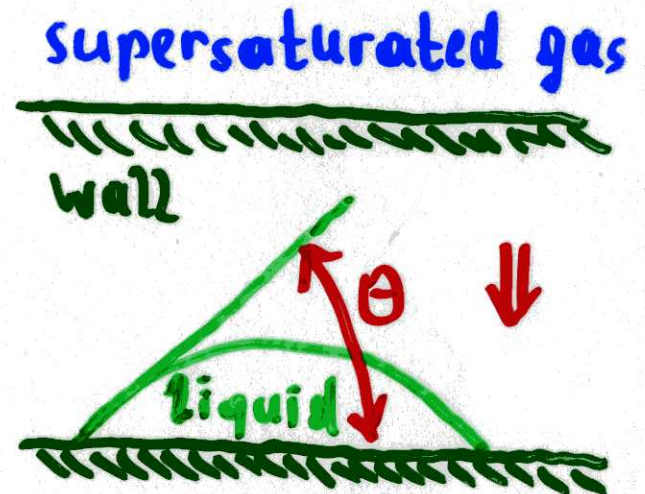
e.g. condensation of a liquid at a wall under INCOMPLETE WETTING conditions

$$\gamma_{wg} - \gamma_{wl} = \gamma_{gl} \cos \theta$$

$\theta =$ contact angle
YOUNG (1805)

- 3 interface tensions
- lower free energy barrier

$$\Delta F_{het}^* = \Delta F_{hom}^* f(\theta)$$



$$f(\theta) = (2 + \cos \theta)(1 - \cos \theta)^2 / 4$$

TURNBULL (1950)

CLASSICAL NUCLEATION THEORY

• estimate free energy barrier ΔF^* to form CRITICAL DROPLET (radius R^*)

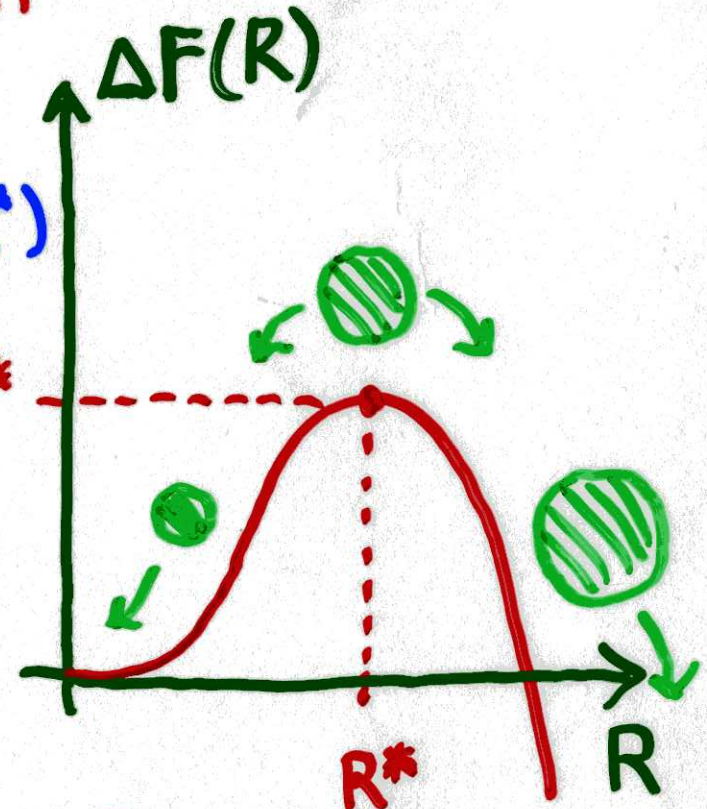
• spherical droplets

• macroscopic description: split $\Delta F(R)$ in BULK and SURFACE terms

$$\Delta F(R) = \Delta g \frac{4\pi R^3}{3} + \gamma_{v2} 4\pi R^2$$

$$\Delta g = -(\rho_l - \rho_v) \delta\mu$$

SAME interfacial free energy as for a FLAT PLANAR INTERFACE



(near coexistence curve) $\delta\mu = \mu - \mu_{\text{coex}}$
chemical potential difference

$$\left. \frac{\partial(\Delta F(R))}{\partial R} \right|_{R^*} = 0 \Rightarrow R^* = \frac{2\gamma_{v2}}{(\rho_l - \rho_v)\delta\mu}, \quad \Delta F^* = \frac{16\pi}{3} \frac{\gamma_{v2}^3}{[(\rho_l - \rho_v)\delta\mu]^2}$$

nucleation rate J^* : # of crit. nuclei/cm³s : $J = \omega^* \exp[-\Delta F^*/k_B T]$

